# WIEN

NIR spectroscopy and chemometrics for ionic liquids: lignocellulosic biomass dissolution



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# Introduction

The dissolution in ionic liquids (IL, [1]) is a new, alternative technology to disrupt the complex fibre network of lignocellulosic biomass at comparatively mild conditions [2]. Its three main compounds – cellulose, hemicelluloses, and lignin – can be separated by simple addition of an anti-solvent; e.g., the addition of water immediately precipitates amorphous cellulose from the IL. Cellulose, hemicelluloses, and lignin are interesting renewable sources for biofuels, chemicals,

# **Dissolution of pure compounds in IL**

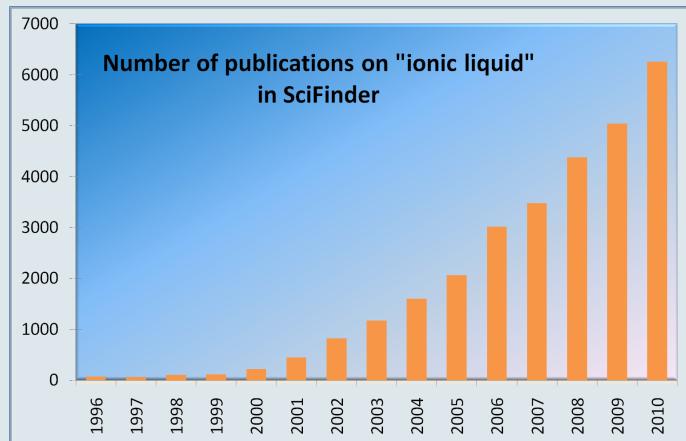
After mixing the pure compounds with the IL, it was necessary to dissolve the particles and agglomerations by ultrasound and temperatures 40-60 °C for 7-24 hours to obtain clear solutions.



and biomaterials. In this preliminary study we investigate the applicability of NIR spectroscopy and chemometrics for quantifying the three main compounds of lignocelluloses in IL.

# **Ionic Liquids**

- consist entirely of ions (salts)
- melting point < 100°C (liquid at room T!)</li>
- negligible vapour pressure
- excellent thermal stability up to 400°C
- non flammable
- electrically conducting, magnetic
- increasing scientific relevance since 2000
- many interesting applications!
- "design" properties by ion combination



### (Bio-)Polymer Processing

• solvent for regenerating cellulose (fibres, films) • enabling technology for manufacturing cellulose • dissolution and separation of lignocellulosic biomass

### Chemical Processing

• reaction media, catalysts, solvents

### Metal Processing

• electroplating (aluminum, chromium) • electropolishing of metal surfaces

### Electrochemistry

• high-performance electrolyte in lithium ion batteries • solvent and electrolyte in solar cells

### Engineering

- non-flammable hydraulic liquids
- high-performing lubricants
- liquid piston and gas compression

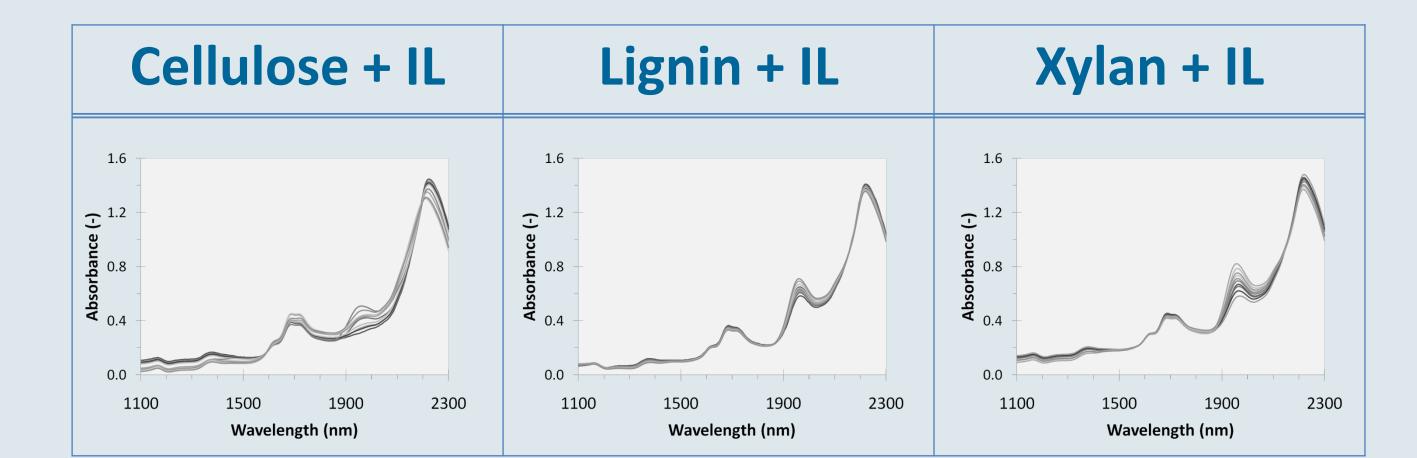
### Separation Processes

- entrainer in extractive distillations (e.g., ethanol/water)
- extraction processes
- purification of gases

### **Functional Fluids**

- antistatic additives, polymer additives
- catalysts

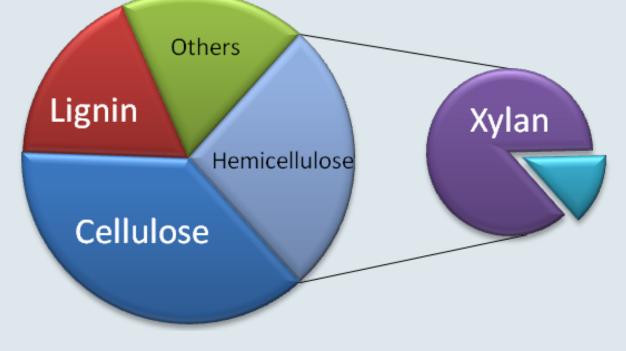
# **NIR absorbance spectra**



• reactive building blocks

# **Lignocellulosic Biomass**

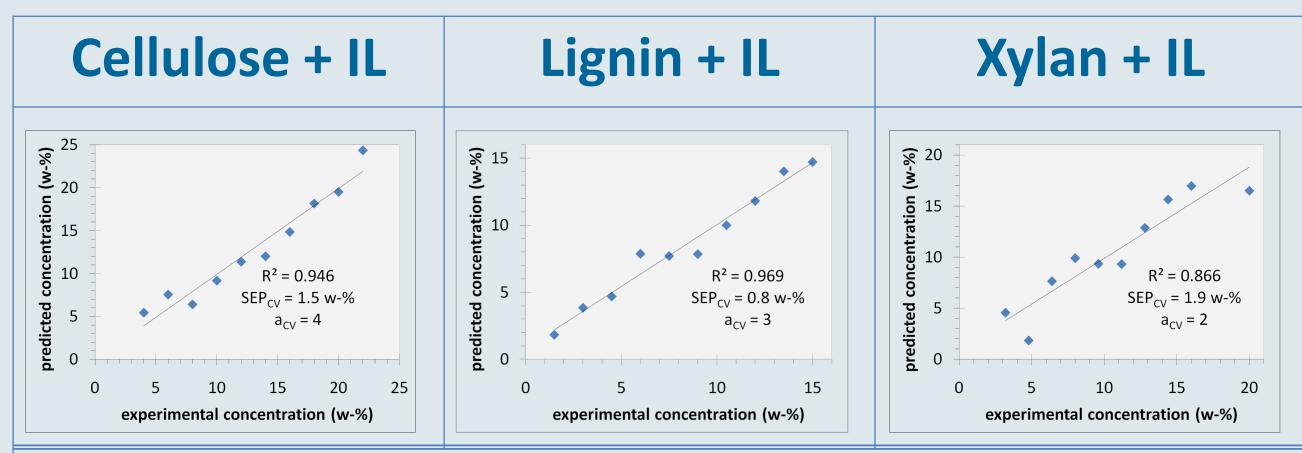
- renewable plant material such as straw or wood
- main components: cellulose, hemicelluloses, lignin (see pie chart, approx. for straw)
- interesting renewable sources for biofuels, chemicals, biomaterials
- **problem**: complex fibre structure is recalcitrant against decomposition
- conventional technology for disruption: high T, high p, aggressive chemicals [3]
- "green" technology: dissolution in "EMIM-OAc", a non-toxic, biodegradable IL [4]



# **Experiments**

- **Ionic liquid**: 1-ethyl-3-methylimidazolium acetate (EMIM-OAc > 98%, >2  $\epsilon/g$ )
- **30 synthetic standards** of IL spiked with (a) <u>cellulose</u> Avicel<sup>®</sup> with 4-22 w-%, (b) <u>lignin</u> "Kraft" with 1.5-15 w-% or (c) hemicellulose <u>xylan</u> from beech wood with 3.2-20 w-%.
- NIR spectrometer (Brimrose Luminar 5030, AOTF), NIR absorbance measured at 1100-2300 nm (5 nm intervals, 241 variables).

# **PLS regression models**



SEP<sub>cv</sub>, standard error of prediction from leave-one-out cross validation a<sub>cv</sub>, number of PLS components PLS-models with leave-one-out CV in Unscrambler [5].

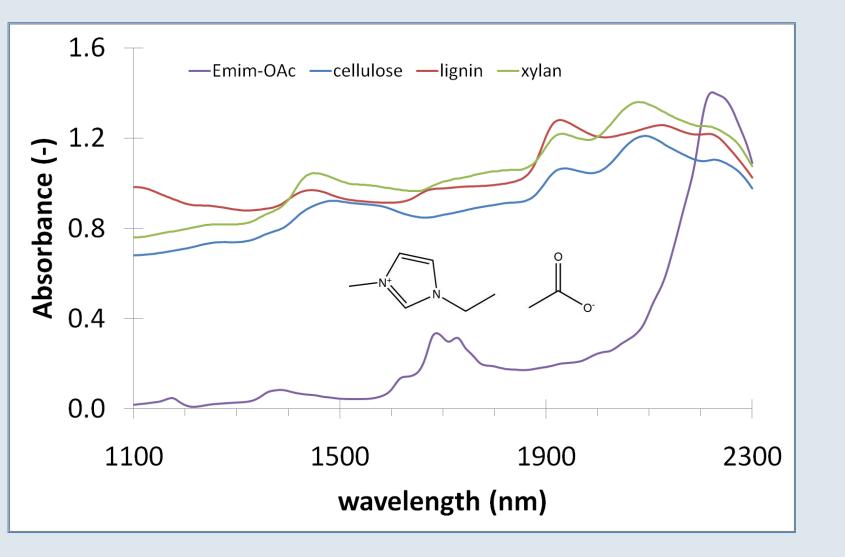
# **Summary and Outlook**

- Empirical models for the concentration of cellulose, lignin, and xylan in ionic liquid EMIM-OAc result in good prediction performance.
- In the next step, experiments will focus on multi-component mixtures of cellulose, lignin, and xylan in the ionic liquid.
- Good reference values for actually dissolved straw components in the IL is a limiting factor in PLS model building for industrial application.



NIR transflectance probe with variable pathlength for liquid samples (up left); NIR diffuse reflectance probe for solid samples (up right).

*Right: NIR spectrum and molecular* structure of the used IL; NIR spectra of pure solid cellulose, lignin, and xylan.



## References

[1] A. Kokorin (ed.): *Ionic liquids: Applications and Perspectives*, InTech, Rijeka, Croatia, 2011. [2] N. Sun, H. Rodriguez, M. Rahman, R.D. Rogers, Where are ionic liquid strategies most suited in the pursuit of chemicals and energy from lignocellulosic biomass? Chem. Commun., 47, 1405 (2011). [3] J.Y. Zhu, X. Pan, R.S. Zalesny Jr., *Pretreatment of woody biomass for biofuel production:* energy efficiency, technologies, and recalcitrance, Appl. Microbiol. Biotechnol., 87, 847 (2010). [4] P. Mäki-Arvela, I. Anugwom, P. Virtanen, R. Sjöholm, J.P. Mikkola, Dissolution of lignocellulosic materials and its constituents using ionic liquids—A review, Ind. Crops Prod., 32, 175 (2010). [5] Software The Unscrambler v 9.0, Camo Process AS, www.camo.no, Oslo, Norway, 2004.

### Acknowledgements

We thank Mag. Roland Kalb for providing the ionic liquid (proionic GmbH, Graz, Austria, www.proionic.com).

# Conferentia Chemometrica 2011 – Sümeg, Hungary, 18.-21.9.2011